Appl. No. 10/815,339 Amdt. Dated September 22, 2006

Reply to Office action of March 23, 2006

IN THE CLAIMS:

1. (Currently Amended) A method of crystallizing a macrolide from a macrolide

starting material comprising the steps of:

a) combining, in a crystallization vessel, a macrolide starting material, a polar

solvent, a hydrocarbon solvent, and water, whereby at least two phases are formed, at

least one of which is a water-rich phase, and wherein the pH of the water-rich phase is at

least about 7,

b) maintaining the combination at for at least 1 hour, whereby a macrolide-rich

phase is formed from which the macrolide crystallizes.

2. (Original) The method of claim 1 further comprising the step of isolating the

macrolide that crystallizes.

3. (Original) The method of claim 1 wherein the combination of step b is maintained

at a temperature of from about -15°C to about 50°C.

4. (Original) The method of claim 3 wherein the combination of step b is maintained

at a temperature of from about -5°C to about 40°C.

5. (Original) The method of claim 4 wherein the combination of step b is maintained

at a temperature of from about -2°C and about 35°C.

6. (Original) The method of claim 1 wherein the combination of step b is maintained

for between 48 and 100 hours.

7. (Original) The method of claim 1 wherein the polar solvent is selected from the

group consisting of alcohols, esters, nitriles and ethers.

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8. (Original) The method of claim 7 wherein the polar solvent is selected from the group consisting of ethyl acetate, acetonitrile, methanol, ethanol, *n*-propanol, *iso*-propanol, *n*-butanol, *iso*-butanol, acetone, diisopropyl ether, dimethyl formamide, and dimethyl acetamide.

- 9. (Original) The method of claim 8 wherein the polar solvent is ethyl acetate.
- 10. (*Original*) The method of claim 1 wherein the hydrocarbon solvent is selected from the group consisting of *n*-hexane, *n*-heptane, octane, *iso*-octane cyclohexane, methylcyclohexane, benzene, toluene, and xylene.
- 11. (Original) The method of claim 10 wherein the hydrocarbon solvent is n-hexane.
- 12. (Original) The method of claim 1 wherein the pH of the water-rich phase is about 8 or higher.
- 13. (Original) The method of claim 1 wherein the water comprises a base selected from NaOH, KOH, Ca(OH)<sub>2</sub>, NH<sub>3</sub>, Et<sub>3</sub>N, diethylamine and pyridine.
- 14. (*Presently Presented*) The method of claim 1 wherein the macrolide is selected from the group consisting of tacrolimus, sirolimus, pimecrolimus, everolimus, and ascomycin.
- 15. (*Original*) A method of crystallizing a macrolide from a macrolide starting material comprising the steps of:
- a) combining a concentrate residue from whole-broth extraction of macrolidecontaining biomatter in a polar solvent with a hydrocarbon solvent, and water, whereby at least two phases are formed, at least one of which is a water-rich phase, and wherein the

pH of the water-rich phase is at least about 7,

b) maintaining the combination at for at least 1 hour, whereby a macrolide-rich phase is formed from which the macrolide crystallizes.

- 16. (*Original*) The method of claim 15 further comprising the step of isolating the macrolide that crystallizes.
- 17. (Original) The method of claim 15 wherein the combination of step b is maintained at a temperature of from about -15°C to about 50°C.
- 18. (Original) The method of claim 17 wherein the combination of step b is maintained at a temperature of from about -5°C to about 40°C.
- 19. (Original) The method of claim 18 wherein the combination of step b is maintained at a temperature of from about -2°C and about 35°C.
- 20. (*Previously Presented*) The method of claim 15 wherein the combination of step b is maintained for between 48 and 100 hours.
- 21. (Original) The method of claim 15 wherein the polar solvent is selected from the group consisting of alcohols, esters, nitriles and ethers.
- 22. (Original) The method of claim 21 wherein the polar solvent is selected from the group consisting of ethyl acetate, acetonitrile, methanol, ethanol, n-propanol, isopropanol, n-butanol, iso-butanol, acetone, diisopropyl ether, dimethyl formamide, and dimethyl acetamide.
- 23. (Original) The method of claim 22 wherein the polar solvent is ethyl acetate.

- 24. (Original) The method of claim 15 wherein the hydrocarbon solvent is selected from the group consisting of *n*-hexane, *n*-heptane, octane, *iso*-octane cyclohexane, methylcyclohexane, benzene, toluene, and xylene.
- 25. (Original) The method of claim 24 wherein the hydrocarbon solvent is *n*-hexane.
- 26. (Original) The method of claim 15 wherein the pH of the water-rich phase is about 8 or higher.
- 27. (Original) The method of claim 15 wherein the water comprises a base selected from NaOH, KOH, Ca(OH)<sub>2</sub>, NH<sub>3</sub>, Et<sub>3</sub>N, diethylamine and pyridine.
- 28. (*Previously Presented*) The method of claim 15 wherein the macrolide is selected from the group consisting of tacrolimus, sirolimus, pimecrolimus, everolimus, and ascomycin.
- 29. (Original) A method of crystallizing a macrolide from a macrolide starting material comprising the steps of:
- a) combining, at a temperature of about 20° to about 25°C, macrolide starting material, ethyl acetate, n-hexane, and a water solution of a base selected from NaOH, KOH, Ca(OH)<sub>2</sub>, NH<sub>3</sub>, (C<sub>2</sub>H<sub>5</sub>)<sub>3</sub>N, diethylamine and pyridine whereby at least two phases are formed, one of which is a water-rich phase, wherein the pH of the water-rich phase is > about 7,
- b) maintaining the combination at a temperature of about 20°C to about 25°C for at least 1 hour, whereby a macrolide-rich phase is formed from which macrolide crystallizes,
  - c) maintaining the combination at a temperature of about 0°C to about 20°C for at

least 1 hour, and

- d) recovering the macrolide that crystallizes.
- 30. (*Previously Presented*) The method of claim 29 wherein the macrolide is selected from the group consisting of tacrolimus, sirolimus, pimecrolimus, everolimus, and ascomycin.
- 31. (Original) The method of claim 29 wherein the pH of the water-rich phase is about 8 or higher.
- 32. (*Original*) A method of crystallizing a macrolide from a macrolide starting material comprising the steps of:
- a) combining, at a temperature of about 20° to about 25°C, a concentrate residue from whole-broth extraction of macrolide-containing biomatter in ethyl acetate, n-hexane, and a water solution of a base selected from NaOH, KOH, Ca(OH)<sub>2</sub>, NH<sub>3</sub>, (C<sub>2</sub>H<sub>5</sub>)<sub>3</sub>N, diethylamine and pyridine whereby at least two phases are formed, one of which is a water-rich phase, wherein the pH of the water-rich phase is > about 7,
- b) maintaining the combination at a temperature of about 20°C to about 25°C for at least 1 hour, whereby a macrolide-rich phase is formed from which macrolide crystallizes,
- c) maintaining the combination at a temperature of about 0°C to about 20°C for at least 1 hour, and
  - d) recovering the macrolide that crystallizes.
- 33. (*Previously Presented*) The method of claim 32 wherein the macrolide is selected from the group consisting of tacrolimus, sirolimus, pimecrolimus, everolimus, and ascomycin.
- 34. (Original) The method of claim 32 wherein the pH of the water-rich phase is about 8 or higher.

- 35. (Original) In a method for crystallizing a macrolide from a macrolide starting material, the step of combining the macrolide starting material, a polar solvent, a hydrocarbon solvent, and water, whereby at least two phases are formed, at least one of which is water rich, wherein the pH of the water-rich phase is at least about 7.
- 36. (Original) In a method for crystallizing a macrolide from a concentrate residue from whole-broth extraction of macrolide-containing biomatter in a polar solvent, the step of combining the macrolide concentrate in the polar solvent, a hydrocarbon solvent, and water, whereby at least two phases are formed, at least one of which is water rich, wherein the pH of the water-rich phase is at least about 7.